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Design and fabrication of nanostructured electrode materials for energy conversion

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KEYWORDS

Photocatalysis, Hydrogen, Solar cell, Visible-light

SHORT SUMMARY

Photocatalysis as a green and sustainable technology has received much attention as its potential solutions for solar energy and environmental aspects. In order to promote the research work of the field and meet the requirements of practical applications, it is necessary to develop high efficiency visible-light-driven photocatalysts, especially semiconductor photocatalysts. This work summarizes our work recently made in this field, focusing on the scientific possibilities offered by different semiconductor photocatalysts for water splitting, organic pollutants degradation, and solar cells. The recent progress is discussed and explained in detail, including the synthesis, characterization, and applications of visible light-driven photocatalysts. Finally, we conclud by summarizing both findings and perspectives.

EXTENDED ABSTRACT

Introduction

Green and efficient energy technologies are crucial where nanotechnology could assist in the paradigm shift from fossil fuels to renewable energy resources [1]. Hydrogen gas has been investigated as an outstanding substitute to the traditional fossil fuels because of its clean combustion process vielding water vapor, being made of water, the main resource on the earth, and the high efficiency of its evolution from and conversion into electricity [2,3]. Due to their outstanding prospective to employ clean, inexhaustible, and permanently accessible solar energy in vital energy conversions [4], semiconductor-assisted photocatalysis has been extensively investigated for about half a century [5,6]. Since the pioneering work of Fujishima and Honda [7], photocatalytic water splitting became one of the top research interests as a green renewable source of energy [8], and TiO2 became one of the most widely utilized photocatalysts owing to its perfect photocatalytic behavior. non-toxicity, long durability, availability, low cost, photostability, and chemical stability, see Figure 1 [9]. However, its rapid e-h recombination rate and large bandgap energy are the major disadvantages of TiO₂ that severely affect its photocatalytic performance and limit its functionality to the UV region, hindering its practical application [10]. Therefore, as shown in Figures 2,3, it still remains a challenge to develop novel, robust, non-toxic, low-cost, and efficient visible-light-driven photocatalysts with high lightharvesting efficiency and abundant catalytically active sites [11].

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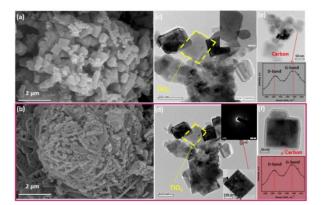


Figure 1 SEM and TEM of TiO₂ mesocrystals

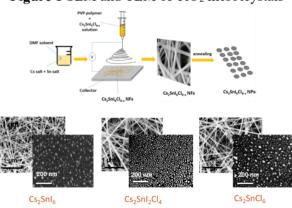


Figure 2 Double perovskites (Cs₂SnI₆Cl_{6-x}) nanofiber

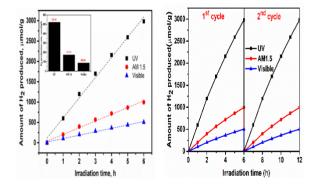


Figure 3 Photocatalytic hydrogen production performance of Bi₂(CrO₄)₃

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